wR = 0.061; all non-H atoms anisotropic, H atoms were included using a riding model [C—H 0.96 Å, U(H) = 0.08 Å<sup>2</sup>, except for: U(H) = 0.12 Å<sup>2</sup> for the methyl H of C(9) and C(14); U(H) = 0.15 Å<sup>2</sup> for the methyl H of C(5) and the olefinic H of C(4)]. 155 parameters were refined, S = 1.63, weighting scheme  $w^{-1} = \sigma^2(F) + 0.0005F^2$  which led to a featureless analysis of variance in terms of  $\sin\theta$  and  $F_o$ , max.  $\Delta/\sigma = 0.001$ , max. and min. height in final  $\Delta\rho$  map 0.16 and -0.16 e Å<sup>-3</sup> respectively. Atomic scattering factors from International Tables for X-ray Crystallography (1974).



Atomic parameters are given in Table 1, selected bond distances and angles in Table 2.\* Fig. 1 shows a thermal-ellipsoid plot with atom numbering.

\* Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52059 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. **Related literature.** The synthesis of the title compound *via* an intramolecular ene reaction has been published (Tietze & Beifuss, 1986). Experimental details of the synthesis and spectroscopic data of the compound will be published (Tietze, Beifuss & Ruther, 1989). For intramolecular ene reactions, see Tietze & Beifuss (1988).

We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for financial support.

#### References

CLEGG, W. (1981). Acta Cryst. A37, 22-28.

- International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht).
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England. Modified by the author.
- SHELDRICK, G. M. (1985). SHELXS86. In Crystallographic Computing 3, edited by G. M. SHELDRICK, C. KRÜGER & R. GOD-DARD, pp. 175–189. Oxford Univ. Press.
- TIETZE, L.-F. & BEIFUSS, U. (1986) Tetrahedron Lett. 27, 1767– 1770.
- TIETZE, L.-F. & BEIFUSS, U. (1988). Justus Liebigs Ann. Chem. 321-329.
- TIETZE, L.-F., BEIFUSS, U. & RUTHER, M. (1989). J. Org. Chem. 54. In the press.

Acta Cryst. (1989). C45, 1834–1836

## Structure of a 3,4-Dihydro-2H-pyran Derivative

By Jochen Antel and George M. Sheldrick

Institut für Anorganische Chemie der Universität, Tammannstraße 4, D-3400 Göttingen, Federal Republic of Germany

### AND UWE HARTFIEL AND LUTZ-F. TIETZE

Institut für Organische Chemie der Universität, Tammannstraße 2, D-3400 Göttingen, Federal Republic of Germany

(Received 19 May 1989; accepted 26 June 1989)

Abstract. (2RS, 3RS, 4RS)- $(\pm)$ -3-Acetoxy-2-ethoxy-4methoxycarbonylamino-3,4-dihydro-2*H*-pyran-5carboxylic acid methyl ester, C<sub>13</sub>H<sub>19</sub>NO<sub>8</sub>,  $M_r =$ 317·30, monoclinic,  $P2_1/n$ ,  $a = 11\cdot839$  (2), b =8·767 (1),  $c = 15\cdot293$  (4) Å,  $\beta = 98\cdot84$  (2)°, V =1568·36 Å<sup>3</sup>, Z = 4,  $D_x = 1\cdot344$  Mg m<sup>-3</sup>,  $\lambda$ (Mo K $\alpha$ ) = 0·71069 Å,  $\mu = 0\cdot11$  mm<sup>-1</sup>, F(000) = 672, T = 298 K, R = 0.060 for 1906 observed reflections. The structure was investigated to determine the relative configuration, which could not be established unambiguously by NMR. The dihydro-2*H*-pyran ring adopts a half-chair conformation.

**Experimental.** (I): The crystal size was  $0.4 \times 0.4 \times 0.7$  mm. The intensity data were collected with a Stoe-Siemens four-circle diffractometer using monochromated Mo  $K\alpha$  radiation and a profile-fitting mode involving variable scan width and speed (Clegg, 1981). 4316 reflections were measured,  $2\theta_{\text{max}} = 50^{\circ}$ ,  $h - 14 \rightarrow 14$ ,  $k0 \rightarrow 10$ ,  $l - 7 \rightarrow 18$ . Three check

0108-2701/89/111834-03\$03.00

© 1989 International Union of Crystallography

Table 1.	Atomic	coordinates	$(\times 10^4)$	and	equivalent
isoti	ropic disp	lacement pa	rameters	$(Å^2)$	$\times 10^{3}$ )

	x	v	z	$U_{eq}$
O(6)	5807 (2)	778 (2)	1427 (1)	57 (1)
0(9)	4654 (2)	- 3334 (2)	2182 (1)	57 (1)
O(1)	3363 (2)	22 (3)	4197 (2)	63 (1)
O(8)	3233 (2)	-4107 (2)	2872 (2)	64 (1)
O(3)	5682 (2)	1971 (2)	3332 (1)	51 (1)
N(5)	4202 (2)	373 (3)	2039 (2)	43 (1)
O(7)	4096 (2)	1708 (2)	804 (1)	59 (1)
C(4)	4757 (2)	- 393 (3)	2834 (2)	41 (1)
O(2)	4257 (2)	2234 (3)	4624 (2)	66 (1)
C(3)	5072 (2)	691 (3)	3609 (2)	44 (1)
C(4a)	4795 (3)	936 (3)	1429 (2)	42 (1)
C(5)	3984 (2)	- 1600 (3)	3118 (2)	41 (1)
C(5a)	3899 (2)	-3126(3)	2729 (2)	45 (1)
O(4)	7311 (2)	643 (3)	3657 (2)	75 (1)
C(6)	3379 (3)	- 1293 (4)	3768 (2)	53 (1)
C(3a)	6823 (3)	1750 (4)	3345 (2)	56 (1)
C(2)	4001 (3)	1304 (4)	3911 (2)	53 (1)
C(3b)	7347 (3)	3024 (4)	2914 (3)	81 (1)
C(2a)	3267 (4)	3144 (6)	4799 (3)	112 (2)
C(5b)	4644 (3)	- 4812 (4)	1761 (3)	76 (1)
C(2b)	3533 (5)	4127 (9)	5478 (4)	195 (3)
C(4b)	4625 (3)	2337 (5)	100 (2)	77 (2)

## Table 2. Bond lengths (Å) and angles (°)

O(6) - C(4a)	·206 (4)	O(9) - C(5a)	1.328 (4)
O(9) - C(5b) 1	446 (4)	$O(1) \rightarrow C(6)$	1.328(4)
O(1) - C(2) 1	.459 (4)	O(8) - C(5a)	1.209(4)
O(3) - C(3)	.433 (4)	$O(3) \rightarrow C(3a)$	1.362 (4)
N(5) - C(4) 1	.454 (3)	N(5) - C(4a)	1.345(4)
$\Omega(7) - C(4a)$ 1	+346(3)	O(7) - C(4b)	1.437 (4)
C(4) - C(3)	·521 (4)	$C(4) \rightarrow C(5)$	1.506 (4)
O(2) - C(2)	1.358 (4)	$O(2) \rightarrow C(2a)$	1.476 (5)
C(2) = C(2)	·513 (4)	$C(5) \rightarrow C(5a)$	1.461(4)
C(5) - C(2) = 1	.339 (4)	$O(4) \rightarrow C(3a)$	$1 \cdot 191(4)$
C(3a) - C(3b) = 1	480 (5)	$C(2a) \rightarrow C(2b)$	1.349 (8)
C(5a) $C(5b)$	1 400 (3)	C(2u) $C(2b)$	1 547 (0)
$C(5a) \rightarrow O(9) \rightarrow C(5b)$	116-3 (2)	C(6) - O(1) - C(2)	118-2 (3)
C(3) - O(3) - C(3a)	115.3 (2)	C(4) - N(5) - C(4a)	122.1 (2)
$C(4a) \rightarrow O(7) \rightarrow C(4b)$	115.8 (2)	N(5) - C(4) - C(3)	112.8 (2)
N(5) - C(4) - C(5)	110.4 (2)	C(3)-C(4)-C(5)	107.5 (2)
$C(2) \rightarrow O(2) \rightarrow C(2a)$	112.7 (3)	O(3) - C(3) - C(4)	109.5 (2)
O(3) - C(3) - C(2)	107.2(2)	C(4) - C(3) - C(2)	110.1 (2)
O(6) - C(4a) - N(5)	125.8 (3)	O(6) - C(4a) - O(7)	123.9 (3)
N(5) - C(4a) - O(7)	110.3 (2)	C(4) - C(5) - C(5a)	122.2 (3)
C(4) - C(5) - C(6)	119.4 (3)	C(5a) - C(5) - C(6)	118.4 (3)
O(9) - C(5a) - O(8)	122.5 (3)	O(9) - C(5a) - C(5)	111.8 (3)
O(8) - C(5a) - C(5)	125.7 (3)	O(1) - C(6) - C(5)	126.3 (3)
O(3) - C(3a) - O(4)	122.7(3)	$O(3) \rightarrow C(3a) \rightarrow C(3b)$	111.7 (3)
O(4) - C(3a) - C(3b)	125.6 (3)	O(1) - C(2) - O(2)	106.5 (3)
O(1) - C(2) - C(3)	108.3(2)	O(2) - C(2) - C(3)	111.4 (2)
O(2) - C(2a) - C(2b)	112.8(4)	-(-, -(-) -(-)	- (-)



Fig. 1. Atom-numbering scheme and 50% probability thermalmotion ellipsoids for the title compound.

reflections showed no significant intensity change. 2755 unique reflections measured ( $R_{int} = 0.022$ ), of which 1906 with  $F > 4\sigma(F)$  were used for all calculations (SHELXS86, Sheldrick, 1985; SHELX76, Sheldrick, 1976). Cell constants refined from  $\pm 2\theta$  values of 40 reflections in the range 20-25°. Absorption and extinction corrections were not necessary. The structure was solved by direct methods. Refinement on Fconverged to R = 0.060, wR = 0.071; all non-H atoms anisotropic. H atoms were included using a riding model [ $\tilde{C}$ —H 0.96 Å, U(H) = 0.08 Å<sup>2</sup>, except for methyl protons  $U(H) = 0.10 \text{ Å}^2$ , with U(H) = $0.25 \text{ Å}^2$  for the methyl protons of C(2b)]. 199 parameters were refined, giving S = 2.00, using a weighting scheme  $w^{-1} = \sigma^2(F) + 0.0005F^2$  which led to a featureless analysis of variance in terms of  $\sin\theta$ and  $F_o$ ,  $\Delta/\sigma \le 0.001$ , max. and min. heights in final  $\Delta\rho$  map 0.24 and -0.32 e Å<sup>-3</sup> respectively. Atomic scattering factors were taken from International Tables for X-ray Crystallography (1974). The large thermal motion of the terminal methyl group C(2b)probably contributed to the relatively high R index.

Atomic parameters are given in Table 1, bond distances and angles in Table 2.\* Fig. 1 shows a thermal-ellipsoid plot with atom numbering.



**Related literature.** For the preparation of the compound see Hartfiel (1987). For the preparation of some related compounds see Tietze, Voss, Harms & Sheldrick (1985) and Voss (1987).

We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for financial support.

#### References

CLEGG, W. (1981). Acta Cryst. A37, 22–28. HARTFIEL, U. (1987). Diploma Thesis, Univ. of Göttingen, Federal Republic of Germany.

<sup>\*</sup> Lists of structure factors, hydrogen coordinates and anisotropic vibrational parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52076 (12 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

- International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press, (Present distributor Kluwer Academic Publishers, Dordrecht.)
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination, modified by the author. Univ. of Cambridge, England.
- SHELDRICK, G. M. (1985). SHELXS86. In Crystallographic Computing 3, edited by G. M. SHELDRICK, C. KRÜGER & R. GODDARD, pp. 175-189. Oxford Univ. Press.
- TIETZE, L.-F., VOSS, E., HARMS, K. & SHELDRICK, G. M. (1985). Tetrahedron Lett. 26, 5273.
- Voss, E. (1987). Thesis, Univ. of Göttingen, Federal Republic of Germany.

Acta Cryst. (1989). C45, 1836-1837

# Structure of (3R,6R,7R,11S)-7-Acetoxymethyl-5,8,10-trioxa-1-azatricvclo[4.3.2.0<sup>3,11</sup>]undecan-4-one

## BY KINGA SUWIŃSKA

Institute of Physical Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01 224 Warsaw, Poland

AND IRMA PANFIL, CZESŁAW BEŁŻECKI AND MAREK CHMIELEWSKI

Institute of Organic Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01 224 Warsaw, Poland

(Received 7 March 1989; accepted 1 June 1989)

Abstract.  $C_{10}H_{13}NO_6$ ,  $M_r = 243.2$ , monoclinic,  $P2_1$ , a  $= 5.558 (2), \quad b = 8.119 (4), \quad c = 11.947 (4) \text{ Å}, \quad \beta = 11.947 (4) \text{ Å},$  $98.19 (2)^{\circ}, V = 534 (2) \text{ Å}^3, Z = 2, D_x = 1.514 \text{ g cm}^{-3},$  $\lambda$ (Mo K $\alpha$ ) = 0.71073 Å,  $\mu$  = 1.185 cm<sup>-1</sup>, F(000) = 256, T = 298 K, final R = 0.046 for 920 observed reflections. The two five-membered rings adopt different conformations: envelope and half-chair. The seven-membered ring shows the distorted chair conformation. The shortest non-bonded contact is 3.29 Å for non-H atoms.

Experimental. Crystal of approximate dimensions  $0.28 \times 0.28 \times 0.30$  mm; intensities measured at 298 K on an Enraf-Nonius CAD-4 four-circle diffractometer (Mo  $K\alpha$  radiation, graphite monochromator). Lattice parameters determined by least squares from 20 reflections ( $12 \le 2\theta \le 28^\circ$ ). Total of 1174 reflec- $\theta = 26^{\circ}$ tions up to  $(0 \le h \le 6,$  $0 \leq k \leq 10$ ,  $-14 \le l \le 14$ ) measured in the  $\omega$ -2 $\theta$  scan mode, 920 reflections considered as observed  $[F_o > 1.5\sigma(F_o)]$ . Three reference reflections monitored every hour showed no significant variation in intensity. No absorption or secondary-extinction correction. Structure solved by MULTAN11/82 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982) and refined by full-matrix least squares with the SDP system (B. A. Frenz & Associates Inc., 1985). Weights of each reflection in refinement (on F) calculated from  $w = 1/\sigma^2(F_o)$ ,  $\sigma(F_o)$  being the e.s.d., based on counting statistics, of the observed structure factor. Scattering factors taken from International Tables for X-ray Crystallography (1974). The total number of parameters refined was 153: one scale

factor, position parameters and anisotropic thermal parameters for non-H atoms; no attempt was made to refine the positions [calculated at d(C-H) =0.95 Å] or isotropic thermal parameters (B = 5.0 Å<sup>2</sup>) of the H atoms. Refinement resulted in final values of R = 0.046, wR = 0.049 and S = 3.18; in the last cycle  $(\Delta/\sigma)_{\text{max}} = 0.02$ . Final max. and min.  $\Delta\rho$  were 0.21 and -0.28 e Å<sup>-3</sup>, respectively. All calculations performed on a MicroPDP11/73 computer. The final

Table 1. Fractional atomic coordinates ( $\times$  10<sup>4</sup>) and equivalent isotropic thermal parameters  $(Å^2)$  for non-H atoms

#### E.s.d.'s are in parentheses.

 $B_{eq} = \frac{4}{3} \left[ a^2 B(1,1) + b^2 B(2,2) + c^2 B(3,3) + ab(\cos \gamma) B(1,2) + b^2 B(2,2) + c^2 B(3,3) + b^2 B(2,2) + b^2 B(2,3) + b^2 B$  $ac(\cos\beta)B(1,3) + bc(\cos\alpha)B(2,3)].$ 

	x	y	Z	Bea
C(1)	6743 (7)	9188 (6)	3037 (3)	2.85 (8)
O(2)	8225 (5)	7740*	2927 (2)	3.35 (6)
C(3)	8373 (8)	7438 (6)	1829 (4)	3.57 (9)
O(3)	9739 (7)	6432 (5)	1555 (3)	6.46 (9)
C(4)	6521 (8)	8443 (6)	1072 (3)	3.62 (9)
C(5)	7581 (8)	9748 (7)	362 (3)	4.1 (1)
N(6)	7212 (6)	11351 (5)	881 (3)	3.56 (7)
C(7)	9187 (8)	11937 (6)	1696 (3)	3.70 (9)
O(8)	10055 (4)	10827 (4)	2582 (2)	2.71 (5)
C(9)	8402 (7)	10615 (5)	3399 (3)	2.48 (7)
C(10)	5205 (7)	9435 (6)	1868 (3)	3.34 (9)
O(11)	5115 (5)	11094 (4)	1457 (2)	3.73 (6)
C(12)	10061 (7)	10343 (6)	4516 (3)	2.89 (8)
O(12)	8813 (5)	9580 (4)	5371 (2)	2.83 (5)
C(13)	7069 (7)	10491 (5)	5764 (3)	2.83 (8)
O(13)	6515 (5)	11839 (4)	5423 (2)	3.66 (6)
C(14)	5975 (8)	9573 (7)	6648 (3)	3.74 (9)

#### \* Origin-defining coordinate.

0108-2701/89/111836-02\$03.00

© 1989 International Union of Crystallography